WORKSHEET 1
PERIODIC TABLE REVIEW

Fill in the blanks on the left with the names and symbols listed. You must use all of the names and symbols one time. Many have more than one correct answer, however there is only one complete set of answers.

_________ the symbol for scandium aluminum
_________ the name of a halogen argon
_________ the symbol of the element below sulfur arsenic
_________ the name of the element K bromine
_________ the name of an element that exists as a diatomic molecule gallium
_________ the symbol for copper hydrogen
_________ the symbol of an alkaline earth metal lithium
_________ the name of an actinide mercury
_________ the name of the element As nickel
_________ the symbol of an alkali metal nitrogen
_________ the name of a metal that combines with bromine to give MBr oxygen
_________ the symbol for sodium phosphorus
_________ the name of a transition metal from “Row 6” potassium
_________ the name of the element Ni rhenium
_________ the name of an element whose properties are similar to Al uranium
_________ the symbol for a metalloid xenon
_________ the symbol of an element that combines with calcium to give CaX₂ Ag
_________ the name of the element above arsenic At
_________ the symbol for silver Co
_________ the name of the element H Cs
_________ the symbol of the element to the right of iron Cu
_________ the symbol for platinum Gd
_________ the name of an element that combines with Mg in a 1:1 ratio I
_________ the name of a metal that is a liquid at room temperature Na
_________ the symbol for tungsten Pb
_________ the name of the element Xe Pt
_________ the symbol of a lanthanide Sc
_________ the name of the element just below boron Se
_________ the symbol of a radioactive element Si
_________ the symbol for lead Sr
_________ the name of a noble gas W